

## The Ultimate Chemical Equations Handbook Answers Chapter 6

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### The Ultimate Chemical Equations Handbook Answers Chapter 7

Note: Not all of the reactions will occur. For those that do not, write no reaction. 1. A piece of copper is dropped into a container of water. No reaction 2. Liquid bromine is added to a container of sodium iodide crystals.  $\text{Br (l)} + 2\text{NaI(s)} \rightarrow 2\text{NaBr(s)} + \text{I}_2\text{(s)}$  3. An aluminum strip is immersed in a solution of silver nitrate.

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Re-typed from The Ultimate Chemical Equations Handbook by Hauge and Smith 7. Free elements, no matter how complex the molecule, have an oxidation number (valence or charge) equal to zero. OJ. Fe. T– He < /1 1 ~– L-td. 8. The following are diatomic or polyatomic elements in nature which must be committed to memory.

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### Quiz on all Chemical Reactions - Max Study

The Ultimate Chemical Equations Handbook SnCO tin(II) carbonate sodium hydrogen carbonate NaHCO 1. 2. 3. 4. 5. 6. 8. 9. 10. 11. vanadium(V) oxide H2O dihydrogen monoxide (NH) CO ammonium oxalate 4 224 polonium(VI) thiocyanate Po(SCN)6 tetraphosphorus decaoxicle P O 4 10 12. 13. 15. 16. 17. 18. 19. manganese(VII) oxide copper(II) dihydrogen phosphate

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### KEY to 5 Reaction Prediction - Ipscience.com

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